



(19) Eur pâlech & Patentamt
European Patent Office
Office européen des brevets



(11) EP 0 731 820 B1

(12)

EUROPEAN PATENT SPECIFICATION

(45) Date of publication and mention of the grant of the patent:
22.04.1998 Bulletin 1998/17

(51) Int Cl. 6: C08G 75/23, C08G 65/40

(21) Application number: 95902228.6

(86) International application number:
PCT/GB94/02662

(22) Date of filing: 02.12.1994

(87) International publication number:
WO 95/15351 (08.06.1995 Gazette 1995/24)

(54) AROMATIC POLYSULPHONES

AROMATISCHE POLYSULFONE
POLYSULFONES AROMATIQUES

(84) Designated Contracting States:
AT BE CH DE DK ES FR GB IE IT LI NL PT SE

- LEWIS, David Frank
Northwich CW8 4TF (GB)

(30) Priority: 02.12.1993 GB 9324731

(74) Representative: Atkinson, Peter Birch et al
MARKS & CLERK,
Sussex House,
83-85 Mosley Street
Manchester M2 3LG (GB)

(43) Date of publication of application:
18.09.1996 Bulletin 1996/38

(56) References cited:
EP-A- 0 008 894
GB-A- 2 245 577

EP-A- 0 008 895

(73) Proprietor: NORTH WEST WATER GROUP PLC
Warrington WA5 3LW (GB)

(72) Inventors:

- COLQUHOUN, Howard Matthew
Knutsford, Cheshire WA16 8QN (GB)

EP 0 731 820 B1

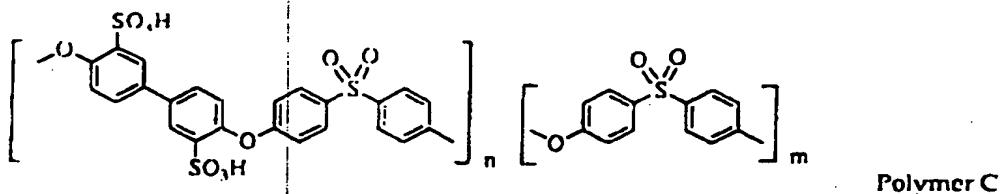
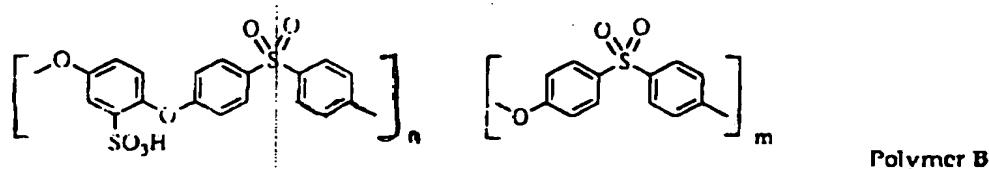
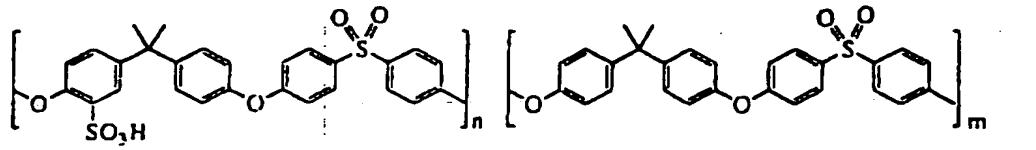
Note: Within nine months from the publication of the mention of the grant of the European patent, any person may give notice to the European Patent Office of opposition to the European patent granted. Notice of opposition shall be filed in a written reasoned statement. It shall not be deemed to have been filed until the opposition fee has been paid. (Art. 99(1) European Patent Convention).

EP 0 731 820 B1

Description

The present invention relates to aromatic polysulphones which are useful particularly, but not exclusively, as membranes for filtration operations.

Composite membranes comprising a thin film of a hydrophilic, sulphonated aromatic polyethersulphone supported on an ultrafiltration membrane are known to have good chemical stability (particularly to extremes of pH and to chlorinated feed waters) and to be usable in reverse osmosis and nanofiltration. Examples of sulphonated polymers which have been used in the past to produce such membranes are shown below as Polymers A, B, and C.

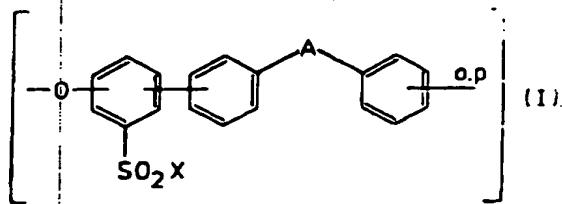


35 Polymer C has particular advantages in that (i) the presence of the biphenyl linkage in place of an isopropylidene or ether linkage provides a material with significantly improved thermo-mechanical and chemical stability, (ii) the precursor (i.e. non-sulphonated) polymer is readily synthesised at high molecular weight, and (iii) sulphonation of this material to give Polymer C proceeds rapidly and cleanly at room temperature.

40 However, using polymers of type C, it has not proved possible to fabricate reverse osmosis membranes with an industrially viable combination of salt rejection and water flux. We have found for example that although composite membranes with good fluxes can be readily obtained from polymers of type C, it is difficult to obtain adequate salt rejection.

45 It is therefore an object of the present invention to provide sulphonated aromatic polysulphones, and membranes fabricated therefrom, which obviate or mitigate the abovementioned disadvantages.

According to a first aspect of the present invention there is provided an aromatic polyethersulphone incorporating the following sub-unit I.



EP 0 731 820 B1

where A is an electron withdrawing group, the designation 'o,p' represents that the bond is at the ortho or para position to the A group, and X is selected from;

-OH;

- 5 -O⁻(Mⁿ⁺)_{1/n} where M is an organic or inorganic cation (other than hydrogen) and n is at least 1;
 -NR₁R₂ where R₁ and R₂ are the same or different and are selected from hydrogen, and alkyl, aryl, aminoalkylene or ammonioalkylene groups; and
 -OR where R₃ is an alkyl or aryl group;

10 either as the only sub-units of the polymer or in combination with comonomer units.
 In the above formula, A is preferably a carbonyl linkage (-CO-) or, more preferably, a sulphonyl linkage (-SO₂⁻). If X is -O⁻(Mⁿ⁺)_{1/n} then the cation may have a valency of one or more. If the valency is greater than 1 then it will be appreciated that the -O⁻ moiety is associated with a fraction 1/n of the cation. M may for example be an alkali metal, alkaline earth metal, or a higher valency metal ion, eg Al³⁺. Preferably M is sodium or potassium.

15 If X is -NR₁R₂ then at least one of R₁ and R₂ is preferably other than hydrogen. If R₁ and/or R₂ is alkyl it preferably has less than 4 carbon atoms. A suitable example of an aminoalkylene group for R₁ and R₂ is -(CH₂)_m-NH₂ where m is less than 4, preferably 2. Correspondingly a suitable ammonioalkylene group is -(CH₂)_m-N⁺R₄R₅R₆ where R₄, R₅ and R₆ are independently hydrogen, alkyl or aryl. If X is -OR₃ and R₃ is alkyl it preferably has less than 4 carbon atoms.

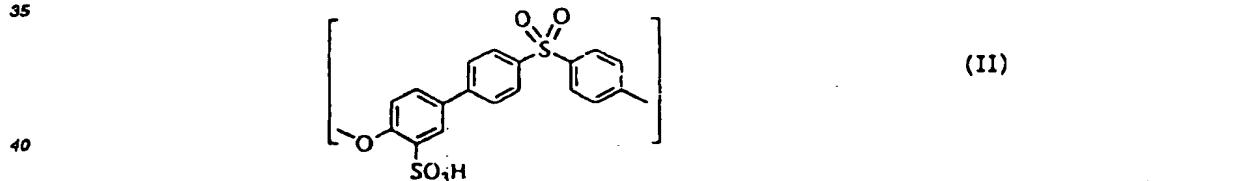
Polymers in accordance with the invention are hydrophilic by virtue of the presence of the -SO₂X groups and are 20 eminently suitable for use as membranes in filtration operations effected in an aqueous environment.

Therefore according to a second aspect of the invention there is provided a porous or non-porous membrane comprising an aromatic polyethersulphone incorporating the sub-unit (I) defined above either alone or in combination with comonomer units.

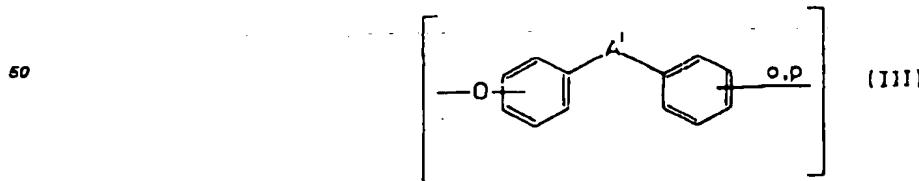
Membranes in accordance with the second aspect of the invention may for example be asymmetric membranes, 25 composite membranes or dense films and may be used in a variety of separation processes including reverse osmosis, electrodialysis, nanofiltration, ultrafiltration, microfiltration, gas separation, pervaporation, haemodialysis and haemo-filtration.

A particularly preferred form of membrane in accordance with the invention is a composite membrane comprising 30 a thin film of the polymer supported on an ultrafiltration membrane. Such composite membranes are particularly suitable as reverse osmosis membranes since they have been surprisingly found not to suffer the problem of poor reverse osmosis performance associated with polymer C (see above).

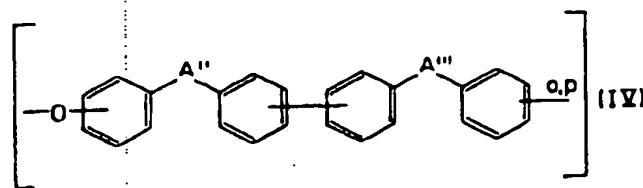
The sub-unit (I) in the polymers of the invention is preferably of the formula (II).



The polymers of the invention may comprise only units (I) (or (II)) or may incorporate comonomer units, eg of the 45 formula (III) or (IV).

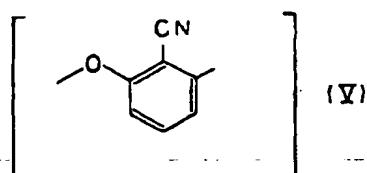


EP 0 731 820 B1

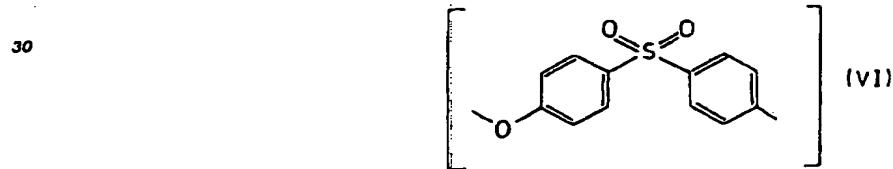


10 In formulae (III) and (IV), A', A'', and A''' represent the same or different electron withdrawing linkages, most preferably carbonyl (-CO-) or sulphone (-SO₂-) linkages. Preferably also all bonds and substituents shown in formulae (III) and (IV) are (on any one aromatic ring) in para relationship to each other.
It is also possible for the comonomer units to be of the formula (V).

15



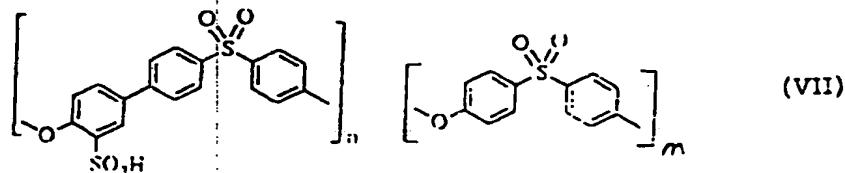
25 The comonomer unit (if present) is most preferably of the formula (VI).



35

If comonomer units are incorporated in the polymer then sub-units (I) may provide 1-99%, more preferably at least 10% eg at least 25%, by mole of the total number of monomer units present in the polymer.
Preferred copolymers in accordance with the invention are of the formula (VII)

40

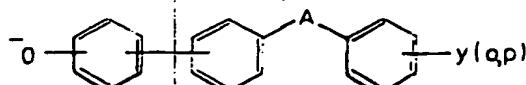


50

where the ratio of m:n is less than 10:1, more preferably less than 4:1, eg 3:1.
The polymers of the invention preferably have a molecular weight of 2000 to 100,000, more preferably 5000 to 50,000. The inherent viscosity of the polymers as measured at a concentration of 1% in N-methylpyrrolidone at 25°C will generally be in the range of 0.3-5.0 dl g⁻¹.

Synthesis of the polymers of the invention for which X = -OH (i.e. the sulphonic acid polymers) may be effected by
(a) polycondensing a monomer containing an anion as represented by formula (VIII)

EP 0 731 820 B1

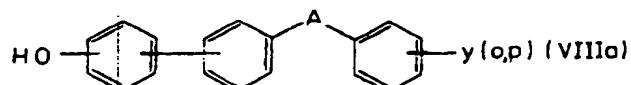


(VIII)

5

where A is an electron withdrawing linkage and Y is a leaving group, preferably selected from halogen or nitro, either alone or in combination with at least one comonomer which is condensable therewith and which gives a substantially non-sulphonatable copolymer unit, said polycondensation or copolycondensation being effected in a solvent for the monomer; and
 (b) sulphonating the resultant polymer.

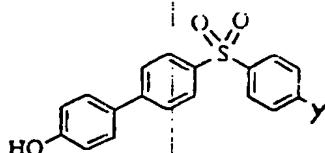
In one embodiment of this process the anion of formula (VIII) is generated *in situ* during the polycondensation or copolycondensation from a monomer of formula (VIIia)



20

where A and Y are as defined above, the polycondensation being effected in the presence of a base. A stoichiometric amount of the base is preferably used. The base may for example be an alkali metal carbonate or an alkali metal fluoride, e.g. potassium fluoride.

It will be appreciated the preferred polymers in accordance with the invention, i.e. those incorporating sub-unit (II) in which X = -OH, may be prepared by polycondensation or copolycondensation of a monomer of formula (IX)



(IX)

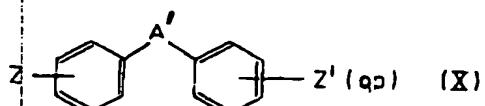
35

Compound IX in which Y = Cl may be named as 4-(4-chlorophenylsulphonyl)-4'-hydroxydiphenyl and can be synthesised by the procedure described in Polymer 1977, Vol 18, April, pages 354 et seq. The analogous compound in which the -SO2- linkage is replaced by a carbonyl linkage (i.e. -CO-) may be synthesised by the procedures described in EP-A-0 232 992 (ICI Plc). Other compounds embraced by formulae (VII) and (IX) may be synthesised by procedures known in the art.

In an alternative embodiment of the process, the anion (VIII) may be obtained by reaction of the corresponding free hydroxy compound with a base (e.g. by reaction with the appropriate alkali metal hydroxide and removal of water) prior to the polycondensation or copolycondensation step.

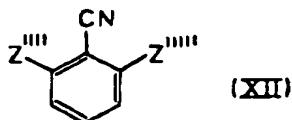
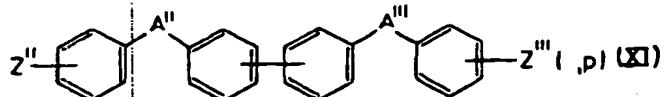
The polycondensation or copolycondensation step will generally be effected at a temperature in the range 150-330°C, preferably 250-300°C.

Comonomers which may preferably be used in step (a) are of the general formula X, XI, and XII



55

EP 0 731 820 B1



where Z , Z' , Z'' , Z''' and Z'''' are the same or different and are selected from hydroxyl groups and leaving groups (preferably halogen atoms (eg chlorine)), and A' , A'' and A''' are as defined above.

15 In the comonomers of formulae X and XI (and correspondingly the copolymer units derived from these comonomers) the aromatic rings which are bonded to the electron withdrawing groups A' , A'' , A''' are non-sulphonatable due to the presence of these latter groups.

20 The solvent may be a dipolar aprotic solvent as commonly used for substitution reactions, eg dimethylformamide, dimethylsulphoxide, N-methylpyrrolidone, sulpholane, dimethyl sulphone and diphenyl sulphone although other solvents which are substantially non-reactive under the reaction conditions may be employed. The preferred solvent is diphenyl sulphone.

The reaction will generally be effected at a temperature in the range 150°-330°, preferably 250°-300° C.

25 The sulphonation step (b) is most conveniently effected using concentrated (95-100%) sulphuric acid most preferably 98% sulphuric acid. Using concentrated sulphuric acid, sulphonation occurs cleanly in high yield without degradation of the polymer. Other sulphonating agents such as oleum, sulphur trioxide, or a sulphur trioxide derivative such as $\text{SO}_3\text{-pyridine}$ or $\text{SO}_3\text{-triethylphosphate}$ may be used.

30 Sulphonation has been found to occur substantially as mono-sulphonation in one ring of the oxybiphenyl residue to give a sulphonated entity as shown for sub-unit (I). The actual position of substitution in that ring will be determined by the position of the other substituents therein. For example, sulphonation of a polymer obtained by condensation of a monomer of formula (IX) yields a polymer sulphonated at the position shown for sub-unit (II).

35 Since sulphonation is specific to the oxybiphenyl residue it will be appreciated that any desired level of sulphonation of the polymer may be obtained by varying the proportion of comonomer units used in step (a).

In an alternative method of synthesis for polymers of formula (I) in which $X = -\text{OH}$, a compound of the formula VII may initially be sulphonated to give the sulphonic acid derivative which can then be condensed, optionally with comonomers as described above, to give the final polymer.

40 The abovedescribed procedures yield the free sulphonic acid form of the polymers (i.e. $X = -\text{OH}$). Such polymers may then be converted into those in which $X = -\text{O}^-\text{M}^+$, $-\text{NR}_1\text{R}_2$ or $-\text{OR}_3$ by procedures known in the art. Thus to obtain polymers in which $X = -\text{O}^-\text{M}^+$, the free sulphonic acid form may be treated with a base incorporating the M^+ cation. Thus if M^{2+} represents an alkali metal, the free sulphonic acid form of the polymer may be treated with an alkali metal hydroxide, eg. NaOH or KOH . Polymers in which $X = -\text{NR}_1\text{R}_2$ may be obtained by conversion of the free sulphonic acid form to the corresponding polyether sulphonyl halide (this conversion may for example be effected with thionyl chloride to obtain the sulphonyl chloride) followed by reaction with the corresponding amine or alcohol. If one of R_1 and R_2 is an ammonioalkylene group then the free sulphonic acid may be reacted with a diamine, the free amine group of which is then converted to an ammonium group. The aforementioned polyether sulphonyl halides are novel intermediates and are therefore provided as a further aspect of the invention.

45 If desired, membranes produced from polymers in accordance with the invention may be over-coated with a cross-linked optionally protected polyol as disclosed in WO 94/17905. The polyol may for example be polyvinylalcohol and may be protected by formaldehyde.

The invention will be further described by way of example only with reference to the following non-limiting examples.

50

Example 1

55 Polymer VII was synthesised in two stages, involving first the polycondensation of monomer (IX) (12.50 mmol, $\text{Y} = \text{Cl}$) with 4,4'-dihydroxydiphenylsulphone (18.75 mmol) and 4,4'-dichlorodiphenylsulphone (18.96 mmol) in the presence of potassium carbonate (25.5 mmol). The reactants were mixed with diphenyl sulphone (56 g) as solvent, blanketed with a slow purge of dry nitrogen, and heated with stirring over 5 hours to 280°C. After 30 minutes at this temperature the viscous reaction mixture was cooled to room temperature and the resulting solid was milled to a fine powder. This was extracted successively with water and then methanol at reflux, and dried to give a 11.0 g of a copolymer.

EP 0 731 820 B1

5 ymer with inherent viscosity in N-methylpyrrolidone (NMP) of 0.66 dl g⁻¹. In a second stage, 3.0 g of this material was dissolved in 50 cm³ of 98% sulphuric acid and the resulting clear, pale yellow solution was allowed to stand for 16 hours. The solution was then added dropwise to stirred deionised water, giving white beads of polymer. After washing 10 hours with water at 60°C, the beads were filtered off and dried at 70°C under vacuum, to give 3.0 g of the sulphonated acid-free with water at 60°C, the beads were filtered off and dried at 70°C under vacuum, to give 3.0 g of the sulphonated

Polymer VII (m=3n), having an inherent viscosity in NMP 0.96 dl g⁻¹.

10 A dry polysulphone ultrafiltration membrane, impregnated with 1,4-butanediol to promote re-wetting of the pores, and having a pure water permeability of 0.56 m³/m².day-bar (6.48×10^{-6} m³/m².s.bar), was coated using a bead-coater with a 0.7% (w/w) solution of Polymer VII (ion exchange capacity 0.92 meq/g) in a solvent comprising methoxyethanol (57%), methanol (38%), sulphuric acid (0.25%), and water (4.75%). The membrane was dried at 80°C. On testing in reverse osmosis (2000 ppm NaCl, 40 bar) the membrane gave a water flux of 78 US gallons/ft².day (GFD) (0.037 l/m².s) and a salt rejection of 95.9%.

Comparative Example 1

15 A dry polysulphone ultrafiltration membrane, as in Example 1, was coated using a bead-coater with a 0.7% (w/w) solution of Polymer C (m = 6n, ion exchange capacity = 0.92 meq/g, inherent viscosity = 0.73 dl g⁻¹) in a solvent comprising methoxyethanol (57%), methanol (38%), sulphuric acid (0.25%), and water (4.75%). The membrane was dried at 80°C. On testing in reverse osmosis (2000 ppm NaCl, 40 bar) the membrane gave a water flux of 135 GFD (0.064 l/m².s) and a salt rejection of 82.9%.

20 It will thus be seen that polymer VII (i.e. one in accordance with the invention) gives superior salt rejection characteristics to one formed from polymer C in an otherwise identical membrane.

Example 2

25 Example 1 was repeated but using proportions of monomers such that after sulphonation the resulting Polymer VII, with m = 1.5n, had an ion exchange capacity of 1.25 meq/g. The inherent viscosity of this polymer in NMP was 1.23 dl g⁻¹. A composite membrane prepared using this material, under the same conditions as those of Example 1, gave a water flux of 4.57 m³/m².day (5.29×10^{-5} m³/m².s) and a salt rejection of 92.7%.

30 On over-coating this membrane with formaldehyde-protected polyvinyl alcohol, as described in our co-pending patent application PCT/GB94/00198 (WO 94/17905), the salt rejection increased to 97.8% and the flux fell to 2.29 m³/m².day (2.65×10^{-5} m³/m².s).

Example 3

35 Example 1 was repeated but using proportions of monomers such that after sulphonation the resulting Polymer VII, with m = 4n, had an ion exchange capacity of 0.74 meq/g. The inherent viscosity of this polymer in NMP was 1.13 dl g⁻¹. A composite membrane prepared using this material, under the same conditions as those of Example 1, gave a water flux of 2.37 m³/m².day (2.74×10^{-5} m³/m².s) and a salt rejection of 96.5%.

40 On over-coating this membrane with formaldehyde-protected polyvinyl alcohol, as described in WO 94/17905 the salt rejection increased to 99.3% and the flux fell to 1.0 m³/m².day (1.16×10^{-6} m³/m².s).

Example 4

This Example demonstrates chlorine tolerance of a membrane based on a polymer of the invention:

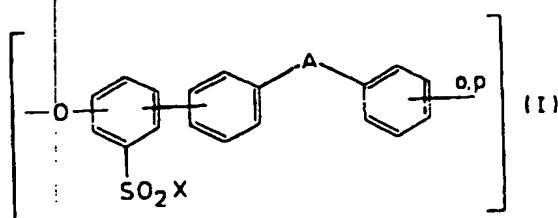
45 A composite membrane prepared as in Example 1 was overcoated with formaldehyde-protected polyvinyl alcohol, as described in WO 94/17905. The salt rejection of the resulting membrane on 500 ppm NaCl at 4 bar was 93.5% and the flux was 0.13 m³/m².day (1.50×10^{-6} m³/m².s). This membrane was run in reverse-osmosis at 60 psi (4.1 bar) with a feed containing 50 ppm of active chlorine at an average pH of 8.0. After a total chlorine exposure of 38,000 ppm-hrs the membrane showed a rejection of 92.7% and a flux of 0.11 m³/m².day (1.27×10^{-6} m³/m².s).

50 Under the same conditions a composite membrane whose separation layer comprised an aromatic polyamide (Dow-Filmtec FT-30) failed almost completely in that its salt rejection fell from 94.4% to less than 20%.

Claims

55 1. An aromatic polyethersulphone incorporating the following sub-unit I.

EP 0 731 820 B1



10

where A is an electron withdrawing group, the designation 'o,p' represents that the bond is at the ortho or para position to the A group, and X is selected from;

-OH;

15

-O(Mⁿ⁺)_{1/n} where M is an organic or inorganic cation (other than hydrogen) and n is at least 1;

-NR₁R₂ where R₁ and R₂ are the same or different and are selected from hydrogen, and alkyl, aryl, aminoalkylene or ammonoalkylene groups; and

-OR₃ where R₃ is an alkyl or aryl group;

20

either as the only sub-units of the polymer or in combination with comonomer units.

2. A polyethersulphone as claimed in claim 1 wherein A is a carbonyl linkage (-CO-).

25

3. A polyethersulphone as claimed in claim 1 wherein A is a sulphonyl linkage (-SO₂-).

4. A polyethersulphone as claimed in any one of claims 1 to 3 wherein X is -OH.

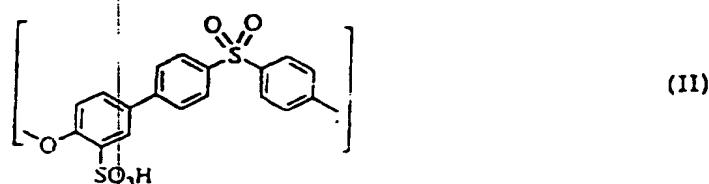
30

5. A polyethersulphone as claimed in any one of claims 1 to 3 wherein X is -O(Mⁿ⁺)_{1/n} wherein M is an alkali metal, alkaline earth metal or a higher valency metal ion.

6. A polyethersulphone as claimed in any one of claims 1 to 3 wherein X is -NR₁R₂ wherein at least one of R₁ and R₂ is other than hydrogen.

35

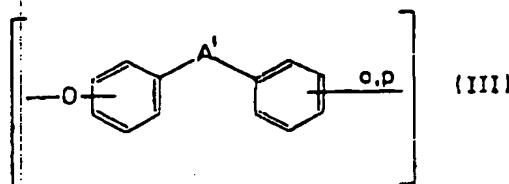
7. A polyethersulphone as claimed in claim 4 wherein sub-unit (I) is of the formula (II)



45

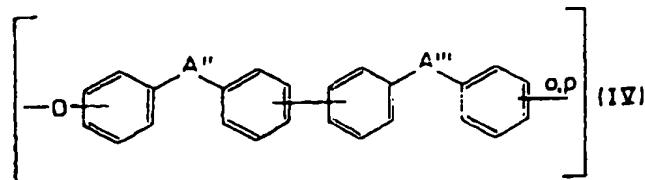
8. A polyethersulphone as claimed in any one of the preceding claims additionally comprising comonomer units of the formula (III) or (IV)

50



55

EP 0 731 820 B1



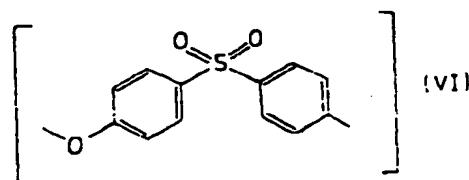
10 In which A', A'' and A''' represent the same or different electron withdrawing linkages.

9. A polyethersulphone as claimed in claim 8 wherein A', A'' and A''' are selected from the group consisting of carbonyl (-CO-) and sulphone (-SO₂) linkages.

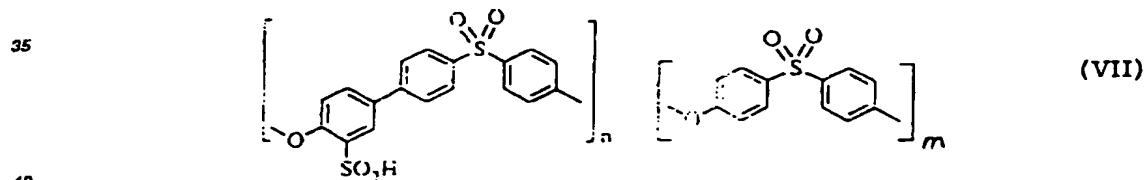
15 10. A polyethersulphone as claimed in claim 8 or 9 wherein all bonds and substituents for units of the formula (III) and (IV) are (on any one aromatic ring) in para relationship to each other.

11. A polyethersulphone as claimed in any one of claims 8 to 10 wherein the comonomer units are of the formula (VI)

20



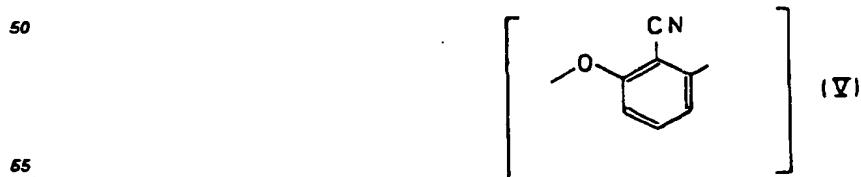
30 12. A polyethersulphone as claimed in claim 11 which is of the formula (VII)



where the ratio of m:n is less than 10:1.

45 13. A polyethersulphone as claimed in claim 12 wherein the ratio of m:n is less than 4:1.

14. A polyethersulphone as claimed in any one of claims 1 to 7 incorporating comonomer units of the formula (V)



EP 0 731 820 B1

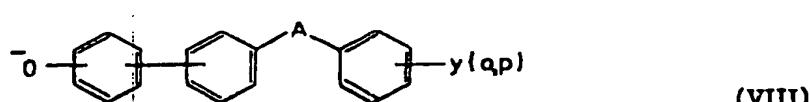
15. A polyethersulphone as claimed in any one of claims 1 to 14 having a molecular weight of 2000 to 100,000.

16. A polyethersulphon as claimed in claim 15 having a molecular weight of 5000 to 50,000.

5 17. A poly thersulphone as claimed in any one of claims 1 to 14 having an inherent viscosity as measured at a concentration of 1% in N-methylpyrrolidone at 25°C is in the range 0.3-5.0 dl g⁻¹.

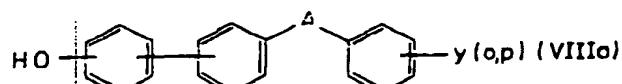
18. A method of synthesising a polymer as claimed in claim 1 and for which X = -OH comprising

10 (a) polycondensing a monomer containing an anion as represented by formula (VIII)



20 where A is an electron withdrawing linkage and Y is a leaving group either alone or in combination with at least one comonomer which is condensable therewith and which gives a substantially non-sulphonatable co-polymer unit, said polycondensation or copolycondensation being effected in a solvent for the monomer; and
 (b) sulphonating the resultant polymer.

25 19. A method as claimed in claim 18 wherein the anion of formula (VIII) is generated in situ during the polycondensation or copolycondensation from a monomer of formula (VIIIa)



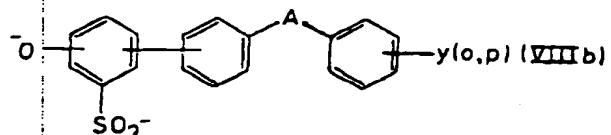
where A and Y are as defined in claim 18, said polycondensation being effected in the presence of a base.

35 20. A process as claimed in claim 19 conducted in the presence of a stoichiometric amount of the base.

21. A process as claimed in claim 19 or 20 wherein the base is an alkali metal carbonate or potassium fluoride.

22. A method of synthesising a polymer as claimed in claim 1 comprising

40 (a) polycondensing a monomer containing an anion as represented by formula (VIIIb)



50 where A is an electron withdrawing linkage and Y is a leaving group either alone or in combination with at least one comonomer which is condensable therewith, said polycondensation or copolycondensation being effected in a solvent for the monomer.

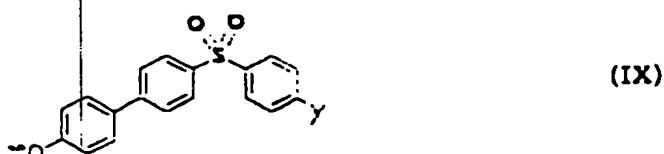
55 23. A process as claimed in any one of claims 18 to 22 wherein A is a carbonyl linkage (-CO-) or a sulphonyl linkage (-SO₂-).

24. A process as claimed in any one of claims 18 to 23 wherein Y is a halogeno- or nitro- substituent.

EP 0 731 820 B1

25. A process as claimed in claim 18 wherein the anion of formula (VIII) is of the formula (IX) in which Y is chlorine.

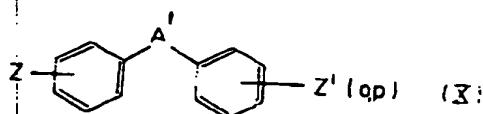
5



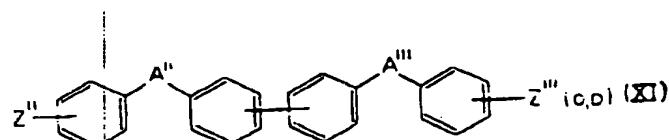
10

26. A process as claimed in any one of claims 18 to 25 which comprises a copolycondensation of the anion of formula (VIII) or (VIIIb) with at least one comonomer of the formula (X), (XI), and (XII)

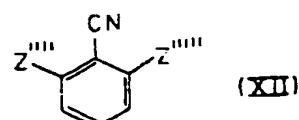
15



20



25



30

where Z, Z', Z'', Z''' and Z'''' are the same or different and are selected from hydroxyl groups and leaving groups (preferably halogen atoms (eg chlorine)), and A', A'' and A''' are as defined hereinbefore.

35

27. A process as claimed in claim 26 wherein said copolycondensation is effected with two comonomers of the formula (X), one of said comonomers of the formula (X) being such that Z and Z' are both leaving groups and the other comonomer of the formula (X) being such that Z and Z' are both hydroxyl groups.

40

28. A process as claimed in claim 27 wherein one of the comonomers of formula (X) is 4,4'-dihydroxydiphenylsulphone and the other comonomer of formula (X) is 4,4'-dichlorodiphenylsulphone.

45

29. A process as claimed in claim 27 or 28 wherein the two comonomers (X) are used in substantially equally molar proportions.

30. A process as claimed in any one of 18 to 29 wherein the solvent is a dipolar aprotic solvent.

50

31. A process as claimed in claim 30 wherein the solvent is selected from dimethylformamide, dimethylsulphoxide, N-methylpyrrolidone, sulpholane, dimethyl sulphone and diphenyl sulphone.

32. A porous or a non-porous membrane comprising a polyethersulphone as claimed in any one of claims 1 to 17.

33. A membrane as claimed in claim 32 which is an asymmetric membrane, composite membrane or dense film.

55

34. A membrane as claimed in claim 32 or 33 adapted for reverse osmosis, electrodialysis, nanofiltration, ultrafiltration, microfiltration, gas separation, pervaporation, haemodialysis or haemofiltration.

EP 0 731 820 B1

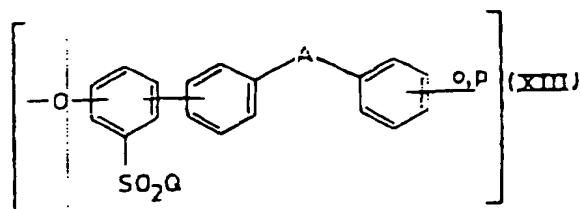
35. A membrane as claimed in claim 32 which comprises a thin film of the polyethersulphonate supported on an ultrafiltration membrane.

36. A membrane as claimed in any one of claims 32 to 35 over-coated with a cross-link and optionally protected polyol.

37. A separation process which comprises the use of a membrane as claimed in any one of claims 32 to 36 as the separating medium.

38. An aromatic polyether sulphonyl halide of the formula (XIII)

10



15

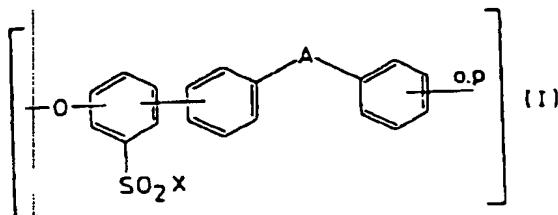
20

where A is as defined in claim 1 and Q is a halogen.

Patentansprüche

25

1. Aromatisches Polyethersulfon, das die folgende Untereinheit I



30

35

enthält, in der A eine elektronenentziehende Gruppe darstellt, die Bezeichnung 'o,p' bedeutet, daß sich die Bindung in Ortho- oder Para-Stellung zur A-Gruppe befindet, und X unter folgenden ausgewählt wird:

-OH;

$-O(M^{n+})_{1/n}$ wobei M ein organisches oder anorganisches Kation (bei dem es sich nicht um Wasserstoff handelt) darstellt und n mindestens 1 beträgt;

$-NR_1R_2$ wobei R_1 und R_2 gleich oder verschieden sind und unter Wasserstoff und Alkyl-, Aryl-, Aminoalkyl- oder Ammonoalkylengruppen ausgewählt werden; und

$-OR_3$ wobei R_3 eine Alkyl- oder Arylgruppe darstellt;

entweder als einzige Untereinheiten des Polymers oder in Verbindung mit Comonomereinheiten.

2. Polyethersulfon nach Anspruch 1, in dem A eine Carbonylbindung (-CO-) darstellt.

3. Polyethersulfon nach Anspruch 1, in dem A eine Sulfonylbindung (-SO₂-) darstellt.

4. Polyethersulfon nach einem der Ansprüche 1 bis 3, in dem X -OH darstellt.

5. Polyethersulfon nach einem der Ansprüche 1 bis 3, in dem X $-O(M^{n+})_{1/n}$ darstellt, wobei M ein Alkalimetall, Erd-